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LETTER TO THE EDITOR

A simple connection between the melting temperature and the glass temperature in a kinetic theory of the glass transition

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Abstract. The relationship of the glass transition temperature T_8 to transport properties has been established. Simple arguments relating transport to distributions of barrier heights, barrier heights to random potential reliefs, and typical potential fluctuations to crystalline potentials allows the establishment of a relationship between T_8 and the melting temperature, T_m , by application of the Lindemann criterion for melting.

A recent Letter to the Editor (Malinovsky and Novikov 1992), referred to hereafter as I, discusses an experimental relationship between the glass temperature T_g and the melting temperature T_m of vitreous and crystalline substances respectively. The relationship is

$$T_* \approx 2T_{\rm m}/3\tag{1}$$

and the proportionality is assumed to result from the excess low-energy density of states (LEDOS) found in glasses. An alternative viewpoint is given here.

The empirical relationship for the viscosity

$$\eta \approx \eta_0 \exp(T_0/T)^2 \tag{2}$$

with η_0 and T_0 system-dependent parameters has been demonstrated (Richert and Baessler (1990), referred to hereafter as II) to hold for a large number of viscous liquids and glasses over many orders of magnitude of η , including the glass transition region, and has been interpreted in II in terms of a random walk theory to result from a Gaussian distribution of barrier heights encountered by a random walker. That η has risen by roughly 13 orders of magnitude compared to its high-temperature liquid value implies a relevant experimental (percolation) relaxation time of

$$t_{\rm exp} = 10^{13} \nu_{\rm ph}^{-1} = 10^{13} 10^{-12} \text{ s} = 10 \text{ s}$$

with $\nu_{\rm ph} \approx 10^{12}$ Hz a nearly universal vibrational frequency (consistent with a relaxation time map emphasized by Angell (1990)). This typical experimental time frame defines the glass temperature. Such a picture has been shown in II (similarly to Hunt (1989)) to lead to (for η given in equation (1))

$$T_{\rm g} = T_0 / \left[\ln(t_{\rm exp} \nu_{\rm ph}) \right]^{1/2} = T_0 / \left[\ln 10^{13} \right]^{1/2} = T_0 / 5. \tag{3}$$

Application of the Lindemann criterion for melting (equivalent to equation (9) of I) gives

$$(1/2)k_{\rm s}r_0^2 = (3/2)k_{\rm B}T_{\rm m} \tag{4}$$

where r_0 is a nearest neighbour distance, k_s is an effective spring constant and k_B is the Boltzmann constant. If the interatomic (interionic) forces are assumed to be primarily Coulombic (ionic and superionic glasses), one has

$$k_{\rm s} \approx \left| {\rm d}F/{\rm d}r \right|_{r_0} = 2e^2/\epsilon r_0^3 \tag{5}$$

(with e the electronic charge and ϵ the dielectric constant) and

$$e^2/\epsilon r_0 = (3/2)k_{\rm B}T_{\rm m} \tag{6}$$

i.e. melting occurs when the thermal energy per particle is of the order of the Madelung energy (numerical Madelung factors of say 1.75 appear in both T_0 and T_m and therefore cancel) per particle. If this same energy is expected to play an important role in transport (viscosity!) then one calculates (Hunt 1990, Henn and Elliott 1992) for a typical barrier height $\langle E_b \rangle$ encountered by a hopping ion (with separation still roughly r_0 and hopping distance of the order of r_0)

$$\langle E_{\rm b} \rangle \approx 2e^2/\epsilon r_0$$
 (7)

(assuming that an ion must pass roughly halfway between other ions to hop over the next barrier). The distribution of such barrier heights, if roughly Gaussian (as assumed in II) will also have a width σ proportional to $\langle E_b \rangle$ (for the distribution to be approximately Gaussian, σ cannot be much larger than $\langle E_b \rangle$ as negative barrier heights make no sense)

$$\sigma \approx 2e^2/\epsilon r_0. \tag{8}$$

But in II T_0 is

$$k_{\rm B}T_{\rm 0} = (2/3)\sigma \approx (4/3)k_{\rm B}T_{\rm m}.$$
 (9)

So, putting together all the proportionalities leads to

$$T_{\rm g} \approx 0.4 T_{\rm m} \tag{10}$$

and the rule $T_{\rm m} \propto T_{\rm g}$ is demonstrated to follow from a kinetic theory of the glass transition.

The proportionality of T_g and T_m is thus demonstrated for ionic and superionic glasses, although in this exceedingly naive calculation the proportionality constant is 40% too small. The restriction to ionic and superionic conductors was made for the purpose of ease of calculation; the relationship between T_g and T_0 is independent of the nature of conduction, i.e. of the nature of the molecular bonding. Whether a similar calculation can be performed for covalently bonded crystals/glasses is not clear, but at least in dipole glasses a large fraction of the bonding as well as of the barrier heights will involve similar electrostatic interactions.

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Reasons for the discrepancy in the numerical constant of proportionality are not difficult to imagine. The Lindemann condition itself is known to be inaccurate, while the rough equality of $\langle E_h \rangle$ and σ already indicates that the approximation of a Gaussian distribution of barrier heights is not quantitatively accurate. Moreover, using r_0 for hopping distances, and for separations in both the crystal and in the glass is not quantitatively accurate. The main point of this work is to clarify the physical origin of the 'glass transition'. The ability of a kinetic theory to predict quantitatively glass transition temperatures in many systems, their dependences on experimentally determined parameters (i.e. T_0) and on the cooling rate as well as the dependence of the width of the 'glass transition' region on the cooling rate could be set against e.g. the observed correlation between $T_{\rm g}$ and $T_{\rm m}$ which was 'explained' in terms of the extra LEDOS or against the correlation between the dynamic heat capacity jump ΔC_{dyn} and the heat of fusion H. But a demonstration that the latter two correlations are also easily understood within a unified conceptual approach based on a purely kinetic interpretation removes the impetus to examine alternative theories. Such a correlation between T_g and T_m being now qualitatively understood, the most important need is to demonstrate the existence (or non-existence) of a proportionality between ΔC_{dyn} and H. Since it has already been demonstrated that ΔC_{dyn} is always a well-defined numerical fraction of the (quasi-) equilibrium heat capacity C_{eq} of a super-cooled liquid, a proportionality between H and C_{eq} remains to be demonstrated.

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